

Coach Ir. Lennert D'ooghe	Supervisor(s) Prof. Dr. Kevin Van Geem Prof. Dr. Vladimir Galvita	Funding
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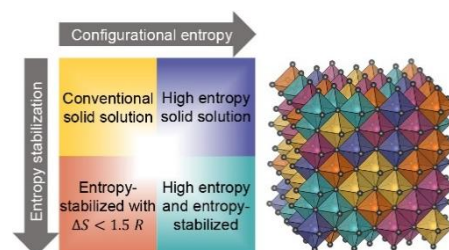
Combined experimental and computational evaluation of high entropy oxide catalysts through Temporal Analysis of Products (TAP)

Aim

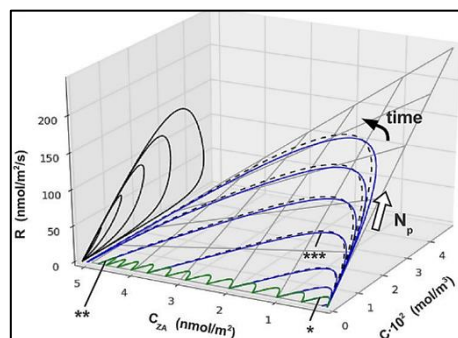
To synthesize and test transition-metal-based high entropy oxide catalysts for CO₂-assisted propane dehydrogenation, leading to structure-performance relations. A particular focus lays on Temporal Analysis of Products (TAP) as experimental and computational kinetic characterization technique.

Justification

Propylene is among the most important chemical building blocks in industry. Its main on-purpose production stems from propane dehydrogenation (PDH). Recent research focusses on introducing CO₂ to this system (CO₂-assisted PDH), holding potential to greatly improve the PDH process while simultaneously providing a way for CO₂ utilization. However, the presence of CO₂ introduces a redox environment for which a stable and selective catalyst remains to be found.



Due to their redox properties, transition-metal-based oxides are catalyst candidates for CO₂-PDH. Especially high entropy oxides have great potential due to geometric and electronic effects between the oxide matrix and the active element. However, little is known about their structure-performance relations for CO₂-PDH, impeding their further development. To advance the knowledge herein, LCT's Temporal Analysis of Products (TAP) reactor setup serves as a unique kinetic characterization tool. Through the combination of molecular beam scattering methodology and the mathematical approach of relaxation theory, mechanistic insights of the gas-solid interaction can be obtained with millisecond time resolution. Combining experiments with the Y-procedure – a mathematical approach developed for TAP – allows to decouple diffusion-reaction phenomena and to extract values for non-steady-state chemical transformation rates.



Top: Sketch of a high entropy oxide. (*J. Am. Chem. Soc.* 145 (2023) 5991–6006).
Bottom: Example of non-steady-state rate/composition space trajectories in TAP. (*Chem. Eng. Sci.* 66 (2011) 6441–6452).

Program

- Literature study on high entropy oxides in view of CO₂-PDH.
- Synthesis of a selection of potential high entropy oxide catalysts for CO₂-PDH.
- Characterization of catalysts through N₂-physisorption, TPR, TPO, TPD, SEM-EDX, etc.
- Activity screening in LCT's step response reactor (PFR).
- TAP experiments on (a selection of) most performant sample(s).
- Extracting diffusion rates and chemical transformation rates from the experimental TAP data, using the Y-procedure.